This article was downloaded by:

On: 14 January 2011

Access details: Access Details: Free Access

Publisher Taylor & Francis

Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-

41 Mortimer Street, London W1T 3JH, UK



Molecular Simulation

Publication details, including instructions for authors and subscription information: http://www.informaworld.com/smpp/title~content=t713644482

On the Possibility of Finding a Suitable Potential Model For Liquid CO,

R. Frattini^a; D. Gazzillo^a; M. Sampoli^b; R. Vallauri^c

^a Dip. Chimica Fisica, University of Venice, Italy ^b Dept. Energetics, University of Florence, Italy ^c Ist. di Elettronica Quantistica, Firenze, Italy

To cite this Article Frattini, R., Gazzillo, D., Sampoli, M. and Vallauri, R.(1989) 'On the Possibility of Finding a Suitable Potential Model For Liquid CO $_{2}$ ', Molecular Simulation, 3: 5, 337 - 342

To link to this Article: DOI: 10.1080/08927028908031385 URL: http://dx.doi.org/10.1080/08927028908031385

PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: http://www.informaworld.com/terms-and-conditions-of-access.pdf

This article may be used for research, teaching and private study purposes. Any substantial or systematic reproduction, re-distribution, re-selling, loan or sub-licensing, systematic supply or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

Preliminary Communication

ON THE POSSIBILITY OF FINDING A SUITABLE POTENTIAL MODEL FOR LIQUID CO₂

R. FRATTINI and D. GAZZILLO

Dip. Chimica Fisica, University of Venice, Italy

M. SAMPOLI

Dept. Energetics, University of Florence, Italy

and

R. VALLAURI

Ist. di Elettronica Quantistica, C.N.R. Via Panciatichi 56/30, 50127 Firenze, Italy

(Received February 1989, accepted February 1989)

The behaviour of the lower harmonic coefficients of the liquid state angular correlation function of CO₂ has been studied using theory and simulation.

KEY WORDS: Liquid CO₂, molecular dynamics, y-expansion

Computer modelling of molecular liquids is still a challenging problem. In particular, liquids composed of linear molecules have been the subject of several investigations [1, 2, 3, 4], and the comparison with X-ray and neutron scattering diffraction data has provided a suitable test of the reliability of the proposed models. In fact, the measurements of the partial structure factors give information on the atom-atom distribution functions, but not directly on the orientational order set up in the liquid phase. Raman spectra are also sensitive to the angular correlations between pairs of molecules: for a realistic interpretation of the experimental data it has been demonstrated that interaction induced effects changing the polarizability of a molecule, have to be taken into account [5], and the dipole-induced-dipole (DID) mechanism is found to give the overwhelming contribution to the scattered intensity. For linear molecules, the effective Raman polarizability can be written in terms of parameters introduced by Ladanyi and Keyes [6], τ_{20} and τ_{22} which are expressed in terms of particular spherical components of the pair radial distribution function, namely;

$$\tau_{20} = 4 \pi \int_0^\infty g_{200}(r) dr/r$$
 (1)

$$\tau_{22} = 4 \pi \int_0^\infty \left[2g_{222}(r) - g_{220}(r) + g_{221}(r) \right] dr/r \tag{2}$$

where the $g_{ll\ m}$ (r) are implicitly defined by [7]

$$g(1, 2) = 4 \pi \sum_{l l' m} g_{l l' m}(r) Y_l^m(\Omega_1) Y_l^{-m}(\Omega_2).$$
 (3)

The τ parameters are sensitive to the change of the spherical component behaviour when different interaction potential models are used [8].

From a computational point of view it would be convenient to rely on some perturbation scheme which allows one to derive the $g_{llm}(r)$ components in terms of a single (and possible simpler) quantity. Such perturbation expansions have been developed [9] and the so called "y-expansion" has been found to reproduce better the computer simulation results for moderately anisotropic systems modelling Br_2 [10]. In essence the y-expansion assumes that the structure of the fluid is determined by an effective two-body potential u_0 (r) obtained by an angular average of the model potential u(1, 2) over the orientation of the molecules: i.e.

$$\exp[-\beta u_0(r)] = \langle \exp[-\beta u(1,2)] \rangle_{\Omega_1,\Omega_2}$$

The centre of mass distribution function is obtained by computer simulation of a reference "monatomic" system at the desired temperature and density where the particles interact through the potential $u_0(r)$. The spherical harmonic components

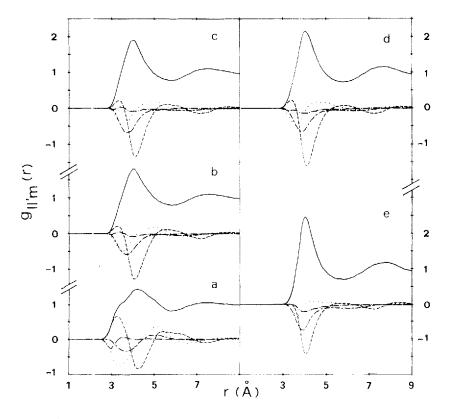


Figure 1 The first spherical harmonic coefficients for the investigated potentials. The labels are: (a) for the potential 2CLJ of reference 1; (b) for 2CLJQ (potential A1 of reference 11); (c) for 3CLJQ [11]; (d) and (e) for the potentials of reference 13 with a quadrupole moment equal to $-4.49 \,\mathrm{DÅ}$ (experimental value) and $-5.8 \,\mathrm{DÅ}$ respectively. Solid lines correspond to g_{000} ; dotted lines to g_{200} ; dashed lines to g_{220} ; dashed-dashed to g_{221} ; dashed-dotted to g_{221} .

 $g_{llm}(r)$ which appear in the expansion (3), are then given in the zero order approximation by [10]:

$$g_{ll'm}(r) = g_0(r) \exp[\beta u_0(r)]^{(0)} g_{ll'm}(r)$$

where $^{(0)}g_{llm}$ represents the spherical harmonic coefficients calculated in the low density limit, i.e. when we use $g(1,2) = \exp[-\beta u(1,2)]$ corresponding to an evaluation performed with only two molecules.

We have analyzed the behaviour of the first harmonic components for different potential models proposed for CO₂, i.e. a two Lennard-Jones centre (2CLJ) model [1], 2CLJ plus a point quadrupole interaction (2CLJQ) [11], a three centre plus quadrupole (3CLJQ) and two models implemented by Böhm et al. [12, 13]. The aim of this investigation is twofold: first to compare the different orientational order induced by different potentials, second to analyze to what extent the perturbation scheme can account for the various spherical harmonics for such an anisotropic system. The probability of occurrence of particular configurations has been calculated in order to have a more complete view of the structure of the liquid and to supplement the study of the atom-atom distribution function already performed in connection with the neutron scattering data [14, 15].

In figure 1 we present the results for various spherical harmonic components obtained by molecular dynamics (MD) simulations of liquid CO_2 at T=222 K and $\varrho=1.15$ g/cm³. The MD runs have been performed with 256 particles in the same

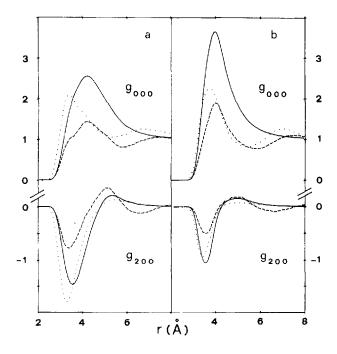


Figure 2 Spherical harmonic components g_{000} and g_{200} for 2CLJ (a) and 3CLJQ (b) potentials, calculated in the low density limit approximation (solid line), in the zero-order y-expansion (dotted line) and by MD simulations (dashed line).

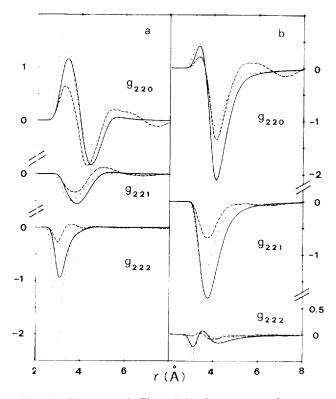


Figure 3 The same as in Figure 2, but for g_{220} , g_{231} and g_{222} .

way as discussed in reference 15 and the averages have been calculated over 6,000 time steps. It is evident that there are large differences among the various potential models, as already expected from the behaviour of the atom-atom distribution functions. A detailed discussion will be presented elsewhere. At present we want to stress that the Böhm's model [13] which reproduces the experimental quadrupole moment, differs negligibly from 3CLJQ, and that for this system the use of distributed charges instead of a point quadrupole has little influence on the overall behaviour. Therefore we shall focus our attention on the comparison between only two models, namely the 2CLJ and 3CLJQ.

For these two potentials we present in figures 2 and 3 the first spherical harmonic coefficients compared with those calculated in the zero order y-expansion and in the low density limit. It appears that the main features are already present in the low density limit results, the major differences being due to a packing effect which reduces the orientational correlations at short distances, but induces evident correlations beyond the first shell of neighbours. The modifications introduced by the packing effect through the centre-centre distribution function in the zero order y-expansion do not improve the behaviour around the second shell distances and lead to better agreement with the MD results at short distances only for the 3CLJQ model. The lack of improvement for the 2CLJ model can be attributed to the very high anisotropy of this potential in the region of the first shell.

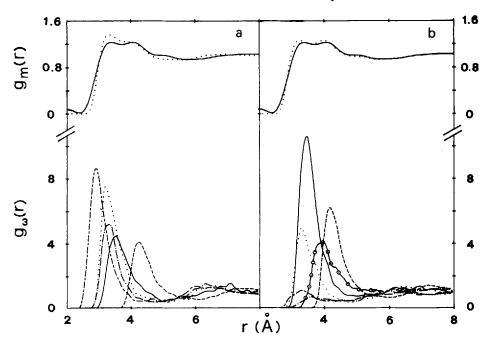


Figure 4 Radial distribution function g_m (r) as probed by neutron scattering (solid lines at the top) together with MD (dotted lines) using the 2CLJ (part a) and 3CLJQ (part b) potentials. The curves at the bottom are the probability distributions for particular configurations: T-shaped (dashed line); Crossed (dashed-dashed line); Parallel (dashed-dotted line); Canted-Parallel at 70° (dotted), 60° (solid) and 50° (solid line with open circles).

The changes of the spherical harmonic coefficients by adopting different potentials are strongly reflected on the values of τ_{20} and τ_{22} , which are found to be -4.52 and 9.88 for 2CLJ respectively, while for 3CLJQ they are -2.35 and 3.69.

Finally, in figure 4 we report the distribution of standard configurations [1, 2, 13] together with the experimental (from neutron scattering) and MD radial distribution function. From the figure one can infer that the better agreement of the 3CLJQ potential is due to the strong reduction of cross and parallel configurations and to the enhancement of T-shaped ones.

References

- [1] K. Singer, A. Taylor and J.V.L. Singer, "Thermodynamic and structural properties of liquids modelled by '2-Lennard-Jones centres' pair potentials." *Mol. Phys.* 33, 1757 (1977).
- [2] W.D. Streett and D.J. Tildesley, "Computer simulations of polyatomic molecules II. Molecular dynamics studies of diatomic liquids with atom-atom and quadrupole-quadrupole potentials." Proc. R. Soc. Lond. A355, 239 (1977)
- [3] C.S. Murthy, K. Singer and R. Vallauri, "Computer simulation of liquid chlorine" Mol. Phys. 49, 803 (1983).; F.P. Ricci, D. Rocca and R. Vallauri, "A Monte Carlo simulation study of liquid Chlorine." Mole. Phys. 60, 1245 (1987).
- [4] D.J. Tildesley and P.A. Madden, "Time correlation functions for a model of liquid carbon disulphide." Mol. Phys. 48, 129 (1983).
- [5] D. Frenkel and J.P. McTague, "Molecular dynamics studies of orientational and collision-induced light scattering in molecular fluids. "J. Chem. Phys. 72, 2801 (1980); "Phenomena induced by

- intermolecular Interactions", G. Birnbaum ed., Plenum Press, New York 1985.
- [6] B.M. Ladanyi and T. Keyes, "The role of local fields and interparticle pair correlations in light scattering by dense fluids. I. Depolarized intensities due to orientational fluctuations." Mol. Phys. 33, 1063 (1977).
- [7] W.B. Streett and D.J. Tildesley, "Computer simulations of polyatomic molecules. I. Monte Carlo studies of hard diatomics." Proc. R. Soc. Lond. A348, 485 (1976).
- [8] A. De Santis and M. Sampoli, "A light scattering study of rotational bands in fluid CO₂", Mol. Phys. 53, 717 (1984).: A De Santis, R. Frattini, M. Sampoli and R. Vallauri, "Raman scattering from CO₂: a computer simulation investigation of effects of polarizability and potential models", Europhys. Lett. 2, 17 (1986)
- [9] J.W. Perram and L.R. White, "A new expansion in the theory of polar luqids." Mol. Phys. 24, 1133 (1972) and "Perturbation theory for the angular correlation function", ibid. 28, 527 (1974); F. Kohler, N. Quirke and J.W. Perram", Perturbation theory with a hard dumbbell reference system. I. Application to liquid nitrogen. "J. Chem. Phys. 71, 4128 (1979); N. Quirke and D.J. Tildesley, "Site superposition approximations for molecular liquids." Mol. Phys. 45, 811 (1982); N. Quirke and D.J. Tildesley, "Perturbation theories of diatomic fluids", J. Phys. Chem. 87, 1972 (1983).
- [10] N. Quirke, J.W. Perram and G. Jacucci, "Perturbation theory for the two-centre Lennard-Jones fluid using a spherically averaged reference potential." Mol. Phys. 39, 1311 (1980).
- [11] K. Singler, C.S. Murthy and I.R. McDonald, "Interaction site models for carbon dioxide," Mol. phys. 44, 135 (1981).
- [12] H.J. Böhm, C. Meissner and R. Ahlrichs, "Molecular dynamics simulation of liquid CH₃F, CHF₃, CH₃Cl, CH₃CN, CO₂ and CS₂ with new pair potentials," Mol. Phys. 53, 651 (1984); H.J. Böhm, R. Ahlrichs, P. Scharf and H. Schiffer, "Intermolecular potentials for CH₄, CH₃F, CHF₃, CH³CL, CH₂Cl₂, CH₃CN and CO₂," J. Chem. Phys. 81, 1389 (1984).
- [13] H.J. Böhm, "The sensitivity of the intermolecular potential to a variation of the potential parameters" Mol. Phys. 56, 375 (1985).
- [14] J.B. van Tricht, H. Frederikze and J. van der Laan, "Neutron diffraction study of liquid carbon dioxide at two thermodynamic states" Mol. Phys. 52, 115 (1984).
- [15] A. De Santis, R. Frattini, D. Gazzillo and M. Sampoli, "The potential model dependence of the neutron radial and partial distribution function for liquid CO2." Mol. Phys. 60, 21 (1987).